UNIVERSAL Chemical Trading GmbH

Version No: 3.1

Safety Data Sheet according to WHS Regulations (Hazardous Chemicals) Amendment 2020 and ADG requirements

Chemwatch Hazard Alert Code: 2

Issue Date: 20/06/2022 Print Date: 30/09/2022 L.GHS.AUS.EN.E

SECTION 1 Identification of the substance / mixture and of the company / undertaking

Product Identifier		
Product name	MOLTEN SULPHUR	
Chemical Name	sulfur, molten	
Synonyms	S; Elemental sulphur; liquid molten sulfur; solid sulfur	
Proper shipping name	SULPHUR, MOLTEN	
Chemical formula	S	
Other means of identification	Not Available	
CAS number	7704-34-9	

Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses

Manufacture of sulfur based chemicals.

Use may require material be molten. Molten or heated material may be compounded, moulded or extruded.

Details of the manufacturer or supplier of the safety data sheet

Registered company name	UNIVERSAL Chemical Trading GmbH	
Address	Waldweg 4 Dollern 21739, Germany	
Telephone	+49-1521-859-2917	
Fax	+49-1521-859-2917	
Website	https://uctr-gmbh.de	
Email	info@uctr-qmbh.de	

Emergency telephone number

Association / Organisation	UNIVERSAL Chemical Trading GmbH	
Emergency telephone		
numbers	+49-1521-859-2917	
Other emergency telephone		
numbers	Not Available	

SECTION 2 Hazards identification

Classification of the substance or mixture

HAZARDOUS CHEMICAL. DANGEROUS GOODS. According to the WHS Regulations and the ADG Code.

Poisons Schedule	Not Applicable Flammable Liquids Category 3, Flammable Solids Category 2, Skin Corrosion/Irritation Category 2, Serious Eye Damage/Eye Irritation Category 2A, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Hazardous to the Aquatic Environment Acute Hazard Category 1	
Classification ^[1]		
Legend:	1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI	

Label elements

Hazard pictogram(s)







Signal word

Warning

Hazard statement(s)

H226	Flammable liquid and vapour.
H228	Flammable solid.
H315	Causes skin irritation.
H319	Causes serious eye irritation.
H335	May cause respiratory irritation.

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H400

Very toxic to aquatic life.

Supplementary statement(s)

Not Applicable

Precautionary statement(s) Prevention

P210	Keep away from heat, hot surfaces, sparks, open flames and other ignition sources. No smoking.	
P271	Use only outdoors or in a well-ventilated area.	

Precautionary statement(s) Response

P370+P378	70+P378 In case of fire: Use alcohol resistant foam or fine spray/water fog to extinguish.	
P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.		

Precautionary statement(s) Storage

P403+P235	Store in a well-ventilated place. Keep cool.	
P405	Store locked up.	

Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

P501

Substances

CAS No	%[weight]	Name
7704-34-9.	100	<u>sulfur</u>
Not Available		tanks on opening may emit gas as
7783-06-4		hydrogen sulfide

Legend:

1. Classified by Chemwatch; 2. Classification drawn from HCIS; 3. Classification drawn from Regulation (EU) No 1272/2008 - Annex VI; 4. Classification drawn from C&L; * EU IOELVs available

Mixtures

See section above for composition of Substances

SECTION 4 First aid measures Description of first aid measures

	Eye Contact	or treat for thermal burn If this product comes in contact with the eyes: Wash out immediately with fresh running water. Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids. Seek medical attention without delay; if pain persists or recurs seek medical attention. Removal of contact lenses after an eye injury should only be undertaken by skilled personnel.
	Skin Contact	If skin contact occurs: Immediately remove all contaminated clothing, including footwear. Flush skin and hair with running water (and soap if available). Seek medical attention in event of irritation. In case of burns: Immediately apply cold water to burn either by immersion or wrapping with saturated clean cloth. DO NOT remove or cut away clothing over burnt areas. DO NOT pull away clothing which has adhered to the skin as this can cause further injury. DO NOT break blister or remove solidified material. Quickly cover wound with dressing or clean cloth to help prevent infection and to ease pain. For large burns, sheets, towels or pillow slips are ideal; leave holes for eyes, nose and mouth. DO NOT apply ointments, oils, butter, etc. to a burn under any circumstances. Water may be given in small quantities if the person is conscious. Alcohol is not to be given under any circumstances. Reassure. Treat for shock by keeping the person warm and in a lying position. Seek medical aid and advise medical personnel in advance of the cause and extent of the injury and the estimated time of arrival of the patient.
	habe to the	 If fumes or combustion products are inhaled remove from contaminated area. Lay patient down. Keep warm and rested. Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.

First aid is not generally required. If in doubt, contact a Poisons Information Centre or a doctor.

Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained.

Indication of any immediate medical attention and special treatment needed

Ingestion

Perform CPR if necessary.

Immediately give a glass of water.

or treat as for a thermal burn

Transport to hospital, or doctor, without delay.

Treat symptomatically.

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SECTION 5 Firefighting measures

Extinguishing media

If fire is in tank, container, replacing lid may smother fire.

- Incipient fires in sulfur storage piles can be frequently smothered by gently shoveling more sulfur, sand, or fine earth on them to exclude all air.
- ▶ For larger fires, water applied as a fine mist is the most useful agent.

For SMALL FIRES:

Dry chemical, CO2, water spray or foam.

For LARGE FIRES:

Water-spray, fog or foam.

Special hazards arising from the substrate or mixture

Fire Incompatibility	Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Advice for firefighters

Fire Fighting	For fires involving sulfur: Do not use solid streams of water; which could create sulfur dust clouds and cause an explosion or move burning sulfur to adjacent areas. Fire will rekindle until mass is cooled below 145 C. Alert Fire Brigade and tell them location and nature of hazard. Wear breathing apparatus plus protective gloves.
Fire/Explosion Hazard	Combustible. Slight fire hazard when exposed to heat or flame. Combustion products include: carbon dioxide (CO2) sulfur oxides (SOx) sulfur dioxide (SO2) other pyrolysis products typical of burning organic material. NOTE: Burns with intense heat. Produces melting, flowing, burning liquid and dense acrid black smoke. Sulfur fires are deep blue at night, with very short flames. Fire is invisible by daylight except for smoke and heat. CARE: Contamination of heated / molten liquid with water may cause violent steam explosion, with scattering of hot contents.
HAZCHEM	1Y

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	Remove all ignition sources. DO NOT touch or walk through spilled material.
Major Spills	 Sulfur dusts form an explosive mixture with air which may be ignited by static electricity. Explosion may be avoided by preventing atmospheres becoming dust-laden by adequate ventilation or by hose-down instead of sweeping. DO NOT touch the spill material Clear area of personnel and move upwind. Alert Fire Brigade and tell them location and nature of hazard.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

Precautions for safe handling	
Safe handling	 Molten sulfur should be maintained at temperatures between 115 deg. minimum, to prevent accumulation of solid sulfur, and 145 deg. DO NOT allow clothing wet with material to stay in contact with skin Avoid all personal contact, including inhalation. Wear protective clothing when risk of overexposure occurs.
Other information	FOR MINOR QUANTITIES: Store in an indoor fireproof cabinet or in a room of noncombustible construction. Provide adequate portable fire-extinguishers in or near the storage area.

Conditions for safe storage, including any incompatibilities

Suitable container	 Glass container is suitable for laboratory quantities For low viscosity materials and solids: Drums and jerricans must be of the non-removable head type. Where a can is to be used as an inner package, the can must have a screwed enclosure.
Storage incompatibility	Sulfur: is both and oxidising agent and a reducing agent when finely divided and dry, forms explosive mixtures with air when molten can generate hydrogen sulfide and carbon disulfide when in contact with some organic materials. is a flammable substance in both the solid and liquid states; the dust is characterised by a very low ignition point of 190 C compared to other combustible dusts - dust clouds are readily ignited by weak frictional sparks if the oxygen content is above 8%. Hydrogen sulfide (H2S): is a highly flammable and reactive gas reacts violently with strong oxidisers, metal oxides, metal dusts and powders, bromine pentafluoride, chlorine trifluoride, chromium trioxide,

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chromyl chloride, dichlorine oxide, nitrogen trichloride, nitryl hypofluorite, oxygen difluoride, perchloryl fluoride, phospham, phosphorus persulfide, silver fulminate, soda-lime, sodium peroxide

- is incompatible with acetaldehyde, chlorine monoxide, chromic acid, chromic anhydride, copper, nitric acid, phenyldiazonium chloride, sodium
- ▶ forms explosive material with benzenediazonium salts
- attacks many metals

Flow or agitation of hydrogen sulfide may generate electrostatic charges due to low conductivity

- Sulfides are incompatible with acids, diazo and azo compounds, halocarbons, isocyanates, aldehydes, alkali metals, nitrides, hydrides, and other strong reducing agents.
- Many reactions of sulfides with these materials generate heat and in many cases hydrogen gas.
- Avoid reaction with oxidising agents

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
German Exposure Standards	hydrogen sulfide	Hydrogen sulphide	10 ppm / 14 mg/m3	21 mg/m3 / 15 ppm	Not Available	Not Available

Emergency Limits

Ingredient	TEEL-1	TEEL-2		TEEL-3
hydrogen sulfide	Not Available	Not Available		Not Available
Ingredient	Original IDLH		Revised IDLH	
	Net Assileble		Not Assoluble	

Ingredient	Original IDLH	Revised IDLH
sulfur	Not Available	Not Available
hydrogen sulfide	100 ppm	Not Available

Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit
sulfur	Е	≤ 0.01 mg/m³
Notes:	Occupational exposure banding is a process of assigning chemicals into adverse health outcomes associated with exposure. The output of this pr to a range of exposure concentrations that are expected to protect worke	ocess is an occupational exposure band (OEB), which corresponds

MATERIAL DATA

Odour Threshold Value for hydrogen sulfide: 0.0011 ppm (detection), 0.0045 ppm (recognition)

NOTE: Detector tubes for hydrogen sulfide, measuring in excess of 0.5 ppm are available commercially.

The TLV-TWA is protective against sudden death, eye irritation, neurasthenic symptoms such as fatigue, headache, dizziness, and irritability, or permanent central nervous system effects that may result from acute, subchronic, or acute exposure to hydrogen sulfide.

No exposure limits set by NOHSC or ACGIH

Other protection

Exposed individuals are reasonably expected to be warned, by smell, that the Exposure Standard is being exceeded.

CAUTION: Vapours may be irritating.

Odour Safety Factor (OSF) is determined to fall into either Class A or B.

Exposure controls

Only open containers in a sheltered but well ventilated area, preferably outdoors. Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls **Appropriate** can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. For engineering controls large scale or continuous use: Spark-free, earthed ventilation system, venting directly to the outside and separate from usual ventilation systems Provide dust collectors with explosion vents Personal protection Safety glasses with side shields Eve and face protection Chemical goggles. See Hand protection below Skin protection NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. The selection of suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in Hands/feet protection advance and has therefore to be checked prior to the application. When handling hot materials wear heat resistant, elbow length gloves. Rubber gloves are not recommended when handling hot objects, materials Protective gloves eg. Leather gloves or gloves with Leather facing Wear physical protective gloves, e.g. leather. Wear safety footwear. **Body protection** See Other protection below

▶ When handling hot or molten liquids, wear trousers or overalls outside of boots, to avoid spills entering boots. Usually handled as molten liquid which requires worker thermal protection and increases hazard of vapour exposure.

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- Overalls
- Eyewash unit.
- Some plastic personal protective equipment (PPE) (e.g. gloves, aprons, overshoes) are not recommended as they may produce static electricity.
- For large scale or continuous use wear tight-weave non-static clothing (no metallic fasteners, cuffs or pockets).

Respiratory protection

- Latridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used

For concentrations exceeding 10 ppm hydrogen sulfide or for unknown concentrations:

- Respirators should be equipped with pressure demand regulators and operated in pressure demand mode only. If airline units are used, a 5-minute egress bottle must also be carried.
- Gas masks or other air-purifying respirators must never be used for H2S, due to the poor warning properties of the gas.
- ▶ When exposure concentrations are unknown and respiratory protection is not used, personal H2S warning devices should be worn.
- These devices should not be relied on to warn of life-threatening concentrations.
- ▶ H2S rapidly fatigues the sense of smell; the rotten egg odour disappears quickly even where high concentrations are present.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Red brown HOT liquid at a temperature above 140 C CARE: Contamination of heated / molten liquid with wate		
Physical state	Liquid	Relative density (Water = 1)	2.07
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	232
pH (as supplied)	Not Applicable	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	114	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	444-445	Molecular weight (g/mol)	32.06
Flash point (°C)	205	Taste	Not Available
Evaporation rate	Slow	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	2.003	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	0.035	Volatile Component (%vol)	Nil @ 38 C.
Vapour pressure (kPa)	0.0146	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (Not Available%)	Not Applicable
Vapour density (Air = 1)	3.64	VOC g/L	Not Available

SECTION 10 Stability and reactivity

	1
Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7
Hazardous decomposition products	See section 5

SECTION 11 Toxicological information

Information on toxicological effects

Evidence shows, or practical experience predicts, that the material produces irritation of the respiratory system, in a substantial number of individuals, following inhalation. In contrast to most organs, the lung is able to respond to a chemical insult by first removing or neutralising the irritant and then repairing the damage.

Inhaled

Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual.

There is no evidence that systemic poisoning results from inhalation of sulfur dust.

Sulfur vapour irritates both the upper and lower respiratory passages and if inhaled may cause coughing, conjunctivitis, nausea, vomiting and chest tightness, bronchitis and in extreme pulmonary oedema (sudden or delayed).

Symptoms of hydrogen sulfide (H2S) exposure may include profuse salivation, nausea, vomiting, diarrhoea, giddiness, headache, vertigo,

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amnesia, palpitations, arrhythmia, weakness, muscle cramps, confusion, sudden collapse, unconsciousness and death due to respiratory paralysis (above 300 ppm). Inhalation of (H2S) at low concentrations causes headache, dizziness and upset stomach Indested sulfur is converted to sulfides in the dastrointestinal tract, and indestion of 10 to 20 grams has caused irritation of the GI tract and renal injury. Individuals with known allergies to sulfide drugs may also have allergic reactions to elemental sulfur. Ingestion The material has NOT been classified by EC Directives or other classification systems as "harmful by ingestion". This is because of the lack of corroborating animal or human evidence. Evidence exists, or practical experience predicts, that the material either produces inflammation of the skin in a substantial number of individuals following direct contact, and/or produces significant inflammation when applied to the healthy intact skin of animals, for up to four hours, such inflammation being present twenty-four hours or more after the end of the exposure period. Skin irritation may also be present after prolonged or repeated exposure; this may result in a form of contact dermatitis (nonallergic). The material may accentuate any pre-Skin Contact existing dermatitis condition Skin contact is not thought to have harmful health effects (as classified under EC Directives); the material may still produce health damage following entry through wounds, lesions or abrasions. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream through, for example, cuts, abrasions, puncture wounds or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected. Evidence exists, or practical experience predicts, that the material may cause eye irritation in a substantial number of individuals and/or may produce significant ocular lesions which are present twenty-four hours or more after instillation into the eye(s) of experimental Eve animals. Repeated or prolonged eye contact may cause inflammation characterised by temporary redness (similar to windburn) of the conjunctiva (conjunctivitis); temporary impairment of vision and/or other transient eye damage/ulceration may occur. Exposure to H2S may produce pain, blurred vision, and irritation. These symptoms are temporary in all but severe cases. Long-term exposure to respiratory irritants may result in disease of the airways involving difficult breathing and related systemic problems. Limited evidence suggests that repeated or long-term occupational exposure may produce cumulative health effects involving organs or biochemical systems. There exists limited evidence that shows that skin contact with the material is capable either of inducing a sensitisation reaction in a significant number of individuals, and/or of producing positive response in experimental animals Chronic Sensitive persons can experience skin irritation from repeated exposure to the sulfur dust. Allergic responses can occur. Chronic low level exposures to hydrogen sulfide may produce headache, fatigue, dizziness, irritability and loss of libido. These symptoms may also result from damage produced by isolated or repeated unmeasured peak high level exposures in healthy persons or those suffering from pre-existing neurological diseases. TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg^[1] Eye: no adverse effect observed (not irritating)^[1] MOLTEN SULPHUR Inhalation(Rat) LC50; >5.43 mg/L4h^[1] Skin: adverse effect observed (irritating)^[1] Skin: no adverse effect observed (not irritating)^[1] Oral (Rat) LD50; >2000 mg/kg^[1] TOXICITY IRRITATION dermal (rat) LD50: >2000 mg/kg^[1] Eye (human): 8 ppm irritant Inhalation(Rat) LC50; >5.43 mg/L4h^[1] Eye: no adverse effect observed (not irritating)^[1] sulfur Skin: adverse effect observed (irritating)^[1] Oral (Rat) LD50; >2000 mg/kg^[1]

hydrogen sulfide

TOXICITY
Inhalation(Mouse) LC50; 316.028 ppm4h^[2]

Legend:

1. Value obtained from Europe ECHA Registered Substances - Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise specified data extracted from RTECS - Register of Toxic Effect of chemical Substances

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Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound.

IRRITATION

Not Available

Acute Toxicity	×	Carcinogenicity	×
Skin Irritation/Corrosion	✓	Reproductivity	×
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	✓
Respiratory or Skin sensitisation	×	STOT - Repeated Exposure	×
Mutagenicity	×	Aspiration Hazard	×

Legend:

X - Data either not available or does not fill the criteria for classification

Data evailable to make classification

Skin: no adverse effect observed (not irritating) $^{\left[1\right]}$

SECTION 12 Ecological information

Toxicity

Toxicity				
	Endpoint	Test Duration (hr)	Species	Value Source
MOLTEN SULPHUR	NOEC(ECx)	504h	Crustacea	>100mg/l 2
	LC50	96h	Fish	>207mg/L 4
	Endpoint	Test Duration (hr)	Species	Value Source
sulfur	NOEC(ECx)	504h	Crustacea	>100mg/l 2

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	LC50	96h	Fish	>207mg/L	4
	Endpoint	Test Duration (hr)	Species	Value	Source
	NOEC(ECx)	3960h	Fish	<0.001mg/L	5
hydrogen sulfide	EC50	48h	Crustacea	0.12mg/l	2
	LC50	96h	Fish	075>0.4mg/l	Not Available
Legend:	Ecotox database	1. IUCLID Toxicity Data 2. Europe ECHA Registe 9 - Aquatic Toxicity Data 5. ECETOC Aquatic Ha ncentration Data 8. Vendor Data			

Very toxic to aquatic organisms.

Do NOT allow product to come in contact with surface waters or to intertidal areas below the mean high water mark. Do not contaminate water when cleaning equipment or disposing of equipment wash-waters.

For hydrogen sulfide:

Environmental fate:

Since hydrogen sulfide exists as a gas at atmospheric pressure, partitioning to the air is likely to occur after environmental releases. However, the compound is also soluble in oil and water, and therefore, may partition as well to surface water, groundwater, or moist soil.

Sulfide ion is very toxic to aquatic life, threshold concentration for fresh or saltwater fish is 0.5ppm. The product therefore is very toxic to aquatic life.

DO NOT discharge into sewer or waterways

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
sulfur	LOW	LOW
hydrogen sulfide	LOW	LOW

Bioaccumulative potential

Ingredient	Bioaccumulation	
sulfur	LOW (LogKOW = 0.229)	
hydrogen sulfide	LOW (LogKOW = 0.229)	

Mobility in soil

Ingredient	Mobility
sulfur	LOW (KOC = 14.3)
hydrogen sulfide	LOW (KOC = 14.3)

SECTION 13 Disposal considerations

Waste treatment methods

Legislation addressing waste disposal requirements may differ by country, state and/ or territory. Each user must refer to laws operating in their area.

Product / Packaging disposal

- DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- Recycle wherever possible.
- Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.

SECTION 14 Transport information

Labels Required



Marine Pollutant



1Y

HAZCHEM

Land transport (ADG)

UN number	2448	2448		
UN proper shipping name	SULPHUR,	SULPHUR, MOLTEN		
Transport based along/an)	Class	4.1		
Transport hazard class(es)	Subrisk	Not Applicable		

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Packing group		
Environmental hazard	Environmentally hazardous	
Special precautions for user	Special provisions Not Applicable Limited quantity 0	
Air transport (ICAO-IATA / DGR		
UN number	2448	
UN proper shipping name	Sulphur, molten	

ICAO/IATA Class ICAO / IATA Subrisk ERG Code	4.1 Not Applicable 3L	
Not Applicable		
Environmentally hazardo	pus	
Special provisions Cargo Only Packing In	structions	Not Applicable Forbidden
		Forbidden Forbidden
	ICAO / IATA Subrisk ERG Code Not Applicable Environmentally hazardo Special provisions Cargo Only Packing In Cargo Only Maximum	ICAO / IATA Subrisk Not Applicable ERG Code 3L Not Applicable Environmentally hazardous

Passenger and Cargo Maximum Qty / Pack

Passenger and Cargo Limited Quantity Packing Instructions

Passenger and Cargo Limited Maximum Qty / Pack

Sea transport (IMDG-Code / GGVSee)

ou manoport (miss ocus) oc				
UN number	2448	2448		
UN proper shipping name	SULPHUR, MOLTEN			
Transport hazard class(es)				
Packing group				
Environmental hazard	Marine Pollutant	Marine Pollutant		
Special precautions for user	EMS Number Special provisions Limited Quantities	F-A, S-H Not Applicable 0		

Forbidden

Forbidden

Forbidden

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
sulfur	Not Available
hydrogen sulfide	Not Available

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
sulfur	Not Available
hydrogen sulfide	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

sulfur is found on the following regulatory lists

German Hazardous Chemical Information System (HCIS) - Hazardous Chemicals German Inventory of Industrial Chemicals (GIIC)

International WHO List of Proposed Occupational Exposure Limit (OEL) Values for Manufactured Nanomaterials (MNMS)

hydrogen sulfide is found on the following regulatory lists

German Hazardous Chemical Information System (HCIS) - Hazardous Chemicals German Standard for the Uniform Scheduling of Medicines and Poisons (SUSMP) - Schedule 7 German Inventory of Industrial Chemicals (GIIC)

National Inventory Status

reactional involutory ocurao	
National Inventory	Status
German - GIIC / German	Yes

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National Inventory	Status
Non-Industrial Use	
Canada - DSL	Yes
Canada - NDSL	No (sulfur; hydrogen sulfide)
China - IECSC	Yes
Europe - EINEC / ELINCS / NLP	Yes
Japan - ENCS	No (sulfur)
Korea - KECI	Yes
New Zealand - NZIoC	Yes
Philippines - PICCS	Yes
USA - TSCA	Yes
Taiwan - TCSI	Yes
Mexico - INSQ	Yes
Vietnam - NCI	Yes
Russia - FBEPH	Yes
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.

SECTION 16 Other information

Revision Date	20/06/2022
Initial Date	16/08/2006

SDS Version Summary

Version	Date of Update	Sections Updated
3.1	20/06/2022	Expiration. Review and Update

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.

Definitions and abbreviations

PC-TWA: Permissible Concentration-Time Weighted Average

PC-STEL: Permissible Concentration-Short Term Exposure Limit

IARC: International Agency for Research on Cancer

ACGIH: American Conference of Governmental Industrial Hygienists

STEL: Short Term Exposure Limit

TEEL: Temporary Emergency Exposure Limit $_{\circ}$

IDLH: Immediately Dangerous to Life or Health Concentrations

ES: Exposure Standard OSF: Odour Safety Factor

NOAEL :No Observed Adverse Effect Level

LOAEL: Lowest Observed Adverse Effect Level

TLV: Threshold Limit Value

LOD: Limit Of Detection

OTV: Odour Threshold Value

BCF: BioConcentration Factors

BEI: Biological Exposure Index

GIIC: German Inventory of Industrial Chemicals

DSL: Domestic Substances List NDSL: Non-Domestic Substances List

IECSC: Inventory of Existing Chemical Substance in China

EINECS: European INventory of Existing Commercial chemical Substances

ELINCS: European List of Notified Chemical Substances

NLP: No-Longer Polymers

ENCS: Existing and New Chemical Substances Inventory

KECI: Korea Existing Chemicals Inventory

NZIoC: New Zealand Inventory of Chemicals

PICCS: Philippine Inventory of Chemicals and Chemical Substances

TSCA: Toxic Substances Control Act

TCSI: Taiwan Chemical Substance Inventory

INSQ: Inventario Nacional de Sustancias Químicas

NCI: National Chemical Inventory

FBEPH: Russian Register of Potentially Hazardous Chemical and Biological Substances

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